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8-CHLORODIBENZ[B,F][1,4]OXAZEPINE-10(11H)-CARBOXYLIC ACID, 2-[3-[2-(FURANYLMETHYL)THIO]-1-OXOPROPYL]HYDRAZIDE (SC-51322): A POTENT PGE₂ ANTAGONIST AND ANALGESIC

E. Ann Hallinan,* Awilda Stapelfeld, Michael A. Savage, Melvin Reichman

> Department of Chemistry Searle 4901 Searle Parkway Skokie, Illinois 60077

Abstract

SC-51322 is the most potent PGE₂ antagonist (pA₂ = 8.1) and analgesic (ED₅₀ = 0.9 mg/kg) that has been seen in this series of N-substituted dibenzoxazepines.

Introduction

Prostaglandin E₂ (PGE₂), a product of arachidonic acid metabolism, is one mediator of the nociceptive process. PGE₂ has been shown to elicit pain and hyperalgesia in humans and to potentiate the action of bradykinin in the transmission of pain.¹ Ferreira and Vane have postulated that the alleviation of pain with NSAIDs (non-steroidal antiinflammatory drugs) occurs by the inhibition of cyclooxygenase's action on arachidonic acid.² In particular, the inhibition of cyclooxygenase prevents the formation of PGE₂.

Sanner and others have observed that SC-19220 (1) is a functional antagonist of PGE₂-elicited contractions in select tissues in vitro;³ while Hammond et al. have shown that antagonists of PGE₂ such as SC-19220 and pinadoline (2) are analgesic.⁴ The rationale of our analgesia program is

based on the hypothesis that PGE₂-induced hyperalgesia occurring in inflamed tissue would be attenuated by selective blockade of PGE₂ receptors of the EP1 subtype in the periphery and in the CNS. Additionally, analgesics based on PGE₂ antagonism would preclude the problems associated with NSAIDs, particularly their gastric side effects.⁵

Chemistry

We sought a clinical replacement for the PGE₂ antagonist-analgesic, pinadoline.⁶ A synthetic effort led to the identification of 3⁷ as a clinical candidate which had PGE₂ antagonism and analgesic activity quite similar to pinadoline. To exploit this lead and to identify a more potent analgesic-PGE₂ antagonist, structural modifications of the alkylsulfonylalkyl moiety of 3 were explored.

Previous research on the alkyl chain of the N-substituted dibenzoxazepines had shown that introduction of heteroaromatic functionality could produce PGE₂ antagonists-analgesics.⁸ The impact of a heteroaromatic group on the alkylthioalkyl moiety had not been investigated. (The oxidation state of sulfur of 3 and its congeners was not crucial for biological activity.) Initial research involved replacement of the terminal methyl of the ethylsulfonylpropionyl functionality of 3 with a furanyl ring. As illustrated in Scheme 1, the syntheses of 4-6 employed techniques reported previously.^{8,9,10}

SCHEME 1

a) NH₂NH₂, EtOH, Δ , 24 h, 89% b) TEA, DCM, 16 h, 79% c) 30% H₂O₂, HOAc, 1 h, rt, 79% d) 30% H₂O₂, HOAc, 24 h, 55*, 22%.

Pharmacology

To determine the effectiveness of 4-6 as analgesics and PGE₂ antagonists, they were tested in the mouse writhing assay¹¹ and PGE₂ antagonism assay.⁸ Analog 6 was found to have in vitro activity comparable to 3; but, it had only marginal analgesic activity and 5 had activity similar to 6. The penultimate precursor of 6, 4 (SC-51322), was the most active PGE₂ antagonist that has been seen in this chemical series. To confirm that the analgesic activity was due to PGE₂ antagonism and not inhibition of cyclooxygenase, SC-51322 was screened by Panlabs in its cyclooxygenase inhibition assay. Also to ensure that SC-51322 was a selective PGE₂

antagonist, it was screened in Panlabs general pharmacology screen which found no other biological activity. 12

Modification of 3 has yielded SC-51322, the most potent PGE₂ antagonist seen in this structural class. SC-51322 will be a powerful tool in further elucidating the role of PGE₂ in the nociceptive process and other PGE₂ mediated disorders.

Table 1: Bioassay Data for PGE₂ Antagonists

No.	R	PGE ₂ Antagonsim Assay in GPI ^a	Mouse Writhing Assay (i.g.c at 30 mg/kg)
	400000000000000000000000000000000000000	pA2 ^b	ED ₅₀ d
2	$(CH_2)_4Cl$	6.2±0.2	9.8
4	~~s~~°	8.1±0.2	0.9 (0.6-1.5)
5	~ § ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~	6.2	5/10
6	~; \$; ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~ ~	6.6	5/10

a. guinea pig ileum. b. pA₂ determined based on the dose ratio at 3 μ M. c. intra gastric. d. The initial screening dose of test compound is 30 mg/kg. Values in parentheses are confidence limits determined at 95% (P<0.05).

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- 10. **4:** Analysis calcd. for $C_{22}H_{20}N_3O_4SCl$ (M.W. 457.94): C, 57.70; H, 4.40; N, 9.18; Cl, 7.74; S, 7.00. Found: C, 57.59; H, 4.36; N, 9.01; Cl, 7.95; S, 7.07. Only the peaks of the major rotamer are reported.^{8,13} ¹H NMR (DMSO-d6.) 9.58 (s, 1H), 8.56 (d, 1H, J = 1 Hz), 7.57 (dd, 1H, J = 0.9, 1.8 Hz),

7.44 (d, 1H, J = 1.8 Hz), 7.20-7.42 (m, 4H), 7.16 (dd, 1H, J = 1.1, 8.2 Hz), 7.06(dt, 1H, J = 1.3, 7.4 Hz), 6.38 (dd, 1H, J = 1.8, 3.2 Hz), 6.27 (dd, 1H, J = 1.8, 3.2 Hz)Hz), 4.84 (s, 2H), 3.77 (s, 2H), 2.63 (t, 2H, J = 7.2 Hz), 2.38 (t, 2H, J = 7.2 Hz). ^BC NMR (DMSO-d6.) 170.2, 155.4, 153.4, 151.4, 142.4, 134.2, 129.3, 128.8, 128.6, 128.3, 127.7, 126.2, 123.2, 122.9, 120.1, 110.5, 107.6, 48.8, 33.4, 27.1, 26.5. 5: Analysis calcd. for C₂₂H₂₀N₃O₅SCl (M.W. 473.93): C, 55.76; H, 4.25; N, 8.87; Cl, 7.48; S, 6.77. Found: C, 55.67; H, 4.30; N, 8.79; Cl, 7.47; S, 6.56. ¹H NMR (DMSO-d6.) 9.73 (s, 1H), 8.58 (s, 1H), 7.68 (s, 1H), 7.15-7.48 (m, 6H), 7.04-7.08 (m, 1H), 6.41-6.47 (m, 2H), 4.85 (s, 2H), 4.25 (d, 1H, J = 4.1 Hz), 4.10(d, 1H, J = 4.1 Hz), 2.97-3.04 (m, 1H), 2.76-2.82 (m, 1H). ¹³C NMR (DMSOd6.) 169.7, 155.4, 153.2, 151.4, 145.1, 143.5, 133.9, 129.2, 128.6, 128.5, 128.4, 127.7, 125.9, 123.3, 122.9, 120.1, 111.0, 110.9, 110.8, 49.5, 48.7, 45.6, 20.8. **6**: Analysis calcd. for C₂₂H₂₀N₃O₆SCl.0.5 H₂O (M.W. 498.95): C, 52.96; H, 4.24; N, 8.42; Cl, 7.11; S, 6.43. Found: C, 52.87; H, 4.03; N, 8.37; Cl, 7.52; S, 6.19. ¹H NMR (DMSO-d6.) 9.78 (s, 1H), 8.61 (s, 1H), 7.72 (dd, 1H, J = 0.8, 1.8Hz), 7.15-7.44 (m, 6H), 7.04-7.08 (m, 1H), 6.53 (dd, 1H, J = 0.8, 3.2 Hz), 6.50 (dd, 1H, J = 1.8, 3.2 Hz, 4.85 (s, 2H), 4.69 (s, 2H), 3.29 (m, 2H), 2.54 (m, 2H). ¹³C NMR (DMSO-d6.) 168.9, 155.5, 153.2, 151.5, 144.2, 142.6, 133.9, 129.3, 128.7, 127.9, 125.9, 123.4, 123.0, 120.2, 112.2, 111.3, 51.6, 48.8, 47.1, 25.3.

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